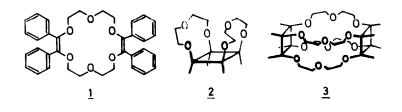
THE PHOTOCHEMISTRY OF 2,3,11,12-TETRAPHENYL-2,3,11,12-DIDEHYDRO-18-CROWN-6

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Summary: Upon UV irradiation the Z - stilbene units of the title compound undergo oxidative cyclization to phenanthrene rings or Z-E - isomerization. New crown ethers containing two phenanthrene rings or one phenanthrene ring and one E- or Zstilbene unit were synthesized.

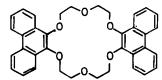
The photochemistry of the unsaturated crown ether 1 [1] appeared to us a possible access to crown derived basket or cage structures like 2 or 3 [2]. We report in this communication on photochemical conversions of 1 which do, however, take a different pathway.



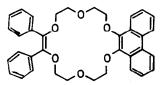
The electronic absorption spectrum of <u>1</u> exhibits absorption bands at 300 nm ( $\varepsilon = 17$  200) and 245 ( $\varepsilon = 9$  800). The variations of the photochemical parameters in our experiments were as follows [3]: irradiation of <u>1</u> in benzene solution (<u>ca</u>. 1.25 x 10<sup>-2</sup> <u>M</u>) at one or both absorption maxima, in the presence or absence of oxidizing agents or triplet sensitizers, and further, irradiation of <u>1</u> or its sodium iodide complex in the solid state. Under all conditions, the only isolated products were the new crown ethers  $\underline{4}$ ,  $\underline{5}$  and  $\underline{6}$  and unreacted  $\underline{1}$ . The product mixtures were separated by low pressure liquid chromatography on a Lobar (Merck) packed silica gel column using  $CH_2Cl_2$  / $CH_3CN$  (9:1) as eluting solvent. Thin layer chromatography revealed the presence of a number of lower molecular weight fragments of  $\underline{1}$ , but species like  $\underline{2}$  and  $\underline{3}$  were not detected in any of the experiments.

When a benzene solution of  $\underline{1}$  was irradiated for 48 h in a quartz vessel in the absence of sensitizers and under rigorous exclusion of oxygen, it remained largely (90 %) unchanged, some fragments and small amounts of  $\underline{4}$ ,  $\underline{5}$  and  $\underline{6}$  being formed.

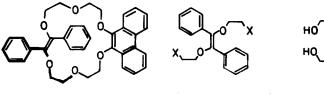
Oxygen as well as ketonic triplet sensitizers serve as oxidants promoting the phenanthrene ring formation by oxidation of the intermediate dihydrophenanthrenes [4]; when <u>1</u> is irradiated in benzene with catalytic amounts of iodine and oxygen bubbling through the solution [5], <u>4</u> is obtained as the sole crown ether in 50-60 % yield after 5 h. Various amounts of <u>4</u>, <u>5</u> and <u>6</u> are obtained if the irradiation is interrupted earlier. The yield of 6 never exceeds 0.5 % .



4, m.p. 245 °C



5, m.p. 218 °C

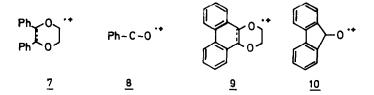


## <u>6</u>, m.p. 228 ℃

11 (a:X=OH; b:X=OTos)

<u>12</u>

The structures of the new crown ethers  $\underline{4}$ ,  $\underline{5}$  and  $\underline{6}$  were confirmed by their spectroscopic properties [6]. In their <sup>1</sup>H - nmr spectra all three species show three typical multiplets (4:2:2) for the phenanthrene protons which are virtually identical in chemical shift and splitting pattern with those of 9,10-dimethoxyphenanthrene [7], and which are the only aromatic proton signals of  $\underline{4}$ . The spectrum of  $\underline{5}$  appears like composed of two halves of each  $\underline{1}$  [1] and  $\underline{4}$  with the phenyl signal as a singlet. The nmr spectrum of  $\underline{6}$  is more complicated: the phenyl protons are split into two multiplets (3:2) as is always noted in E-stilbene derivatives [8], and the methylene protons appear as three multiplets (12:2:2). The mass spectrum of  $\underline{1}$  (m/z 564,M<sup>.+</sup>,100%) contains the key fragments  $\underline{7}$  (m/z 238,70%) and  $\underline{8}$  (m/z 105,97%); the corresponding fragments of  $\underline{4}$  (m/z 560,M<sup>.+</sup>,45%) are  $\underline{9}$  (m/z 236,100%) and  $\underline{10}$  (m/z 180,65%).



The mass spectra of 5 and 6 are identical except for the relative intensities of the fragment ions (m/z 562, M<sup>++</sup>, 52% for 5, 77% for 6); they show all four fragments 7-10 thus indicating the presence of the phenanthrene as well as the stilbene moieties.

In its ir spectrum 5 has the typical C=C stretching frequency of Z-stilbenediol derivatives at 1630 cm<sup>-1</sup> [8], in addition to the phenanthrene 9,10-stretching vibration at 1620 cm<sup>-1</sup> which is shared by all crowns 4-6 and also by 9,10-dimethoxyphenanthrene.

With these data, the structures of <u>4</u> and <u>5</u> are well established. As <u>6</u> is the first example of a crown ether with a trans - double bond its structure was further confirmed by an independent synthesis:  $E-\alpha$ ,  $\beta$ -bis(2-hydroxyethoxy-)stilbene, <u>11a</u> [9], was irradiated in the presence of iodine and oxygen to give the diol <u>12</u> [10]. Cyclization of <u>12</u> with the ditosylate <u>11b</u> (sodium hydride in refluxing THF [11]) gave 12 % of the crown ether <u>6</u> which was in all respects identical with the material obtained photochemically from 1.

## References and notes

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- [2] For similar structures from photochemical cyclization reactions see: H.Krauch, S.Fariad and D.Hess, Chem.Ber. 99, 1880 (1966).
- [3] Irradiations were carried out in a falling thin film quartz reactor using a Philips HPK 125 W low pressure mercury lamp and UV filters as needed.
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- [6] Molecular formulae were established by correct elemental analyses and high resolution mass spectra. <sup>1</sup>H-nmr spectra (in CD<sub>2</sub>Cl<sub>2</sub>, δ values vs. TMS, int.Stand.):
  4: 4.05-4.22,4.39-4.62, symm.m.,16 H; 7.53-7.76,m,8H; 8.13-8.38,m,4H; 8.54-8.79,m,4H.
  - <u>5</u>: 3.80-4.04, symm.m,8H; 4.04-4.21,4.40-4.60, symm.m,8H; 7.20, s,10H; 7.53-7.78, m,4H; 8.15-8.35, m,2H; 8.57-8.80, m,2H.
  - 6:4.00-4.12,m,12H; 4.12-4.40,m,2H; 4.60-4,88,m,2H; 7.01-7.52,m,6 pheny1-H; 7.52-7.85, m, 4 pheny1-H + 4 phenanthrene-H; 8.10-8.38,m,2H; 8.53-8.78,m,2H.
- [7] m.p. 39 <sup>o</sup>C; obtained in 95 % yield from iodine catalyzed photocyclization of E- or Z- $\alpha$ , $\beta$ -dimethoxystilbene [%];<sup>1</sup>H-nmr (CDCl<sub>3</sub>): 4.10,s,6H;7.48-7.78,m,4H;8.13-8.40,m,2H; 8.53-8-80,m,2H.
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- [9] A.Merz, M.Eichner and R.Tomahogh, subsequent paper in this issue.
- [10] 12: m.p. 102 °C, 85 % yield; <sup>1</sup>H-nmr (CDCl<sub>3</sub>): 3.8-4.1,4.1-4.5,symm.m,10H; 7.3-7.7,m,4H; 7.9-8.3,m,2H; 8.4-8.7,m,2H).
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